The Norbornadiene Route to Prostaglandin I₂ and Other Prostaglandins: Preparation and Rearrangement of 7-Substituted Norbornadienes

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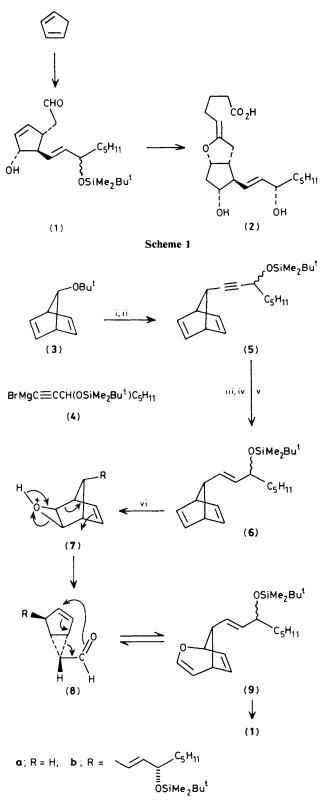
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The cyclopentenylacetaldehyde (1) required for the synthesis of prostacyclin (prostaglandin I_2) and other prostanoids is now readily available from the preparation and rearrangement of 7-substituted norbornadienes.

The first *de novo* synthesis of prostacyclin $(PGI_2)(2)$,¹ a potent inhibitor of blood platelet aggregation, involves eighteen steps from cyclopentadiene and requires the cyclopentenylacetalde-

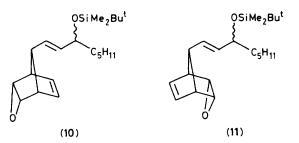
hyde $(1)^2$ as the key intermediate (Scheme 1).¹ We have improved and simplified this synthesis by a study of the preparations and peracid oxidations of 7-substituted nor-



Scheme 2. Reagents: i, HCl, diethyl ether; ii, (4); iii, Bu₄NF or HF; iv, LiAlH₄; v, ClSiMe₂Bu^t; vi, MeCO₃H, Na₂CO₃.

bornadienes whereby the first ten stages of the first synthesis have been replaced by five steps to give the key aldehyde (1) in acceptable yields. This approach also offers a feasible route to other prostanoids with a variety of lower side chains.

Although 7-t-butoxynorbornadiene (3) is known to react



with simple primary alkyl and aryl Grignard reagents³ it did not react with the alkynyl Grignard reagent (4). However, 7chloronorbornadiene generated from 7-t-butoxynorbornadiene (3) by treatment with hydrogen chloride in diethyl ether,⁴ reacted with reagent (4) in tetrahydrofuran in the presence of a catalytic amount of copper(I) chloride to give the 7-alkynylnorbornadiene (5) in 65% yield. That substitution had occurred at C-7 without rearrangement was confirmed by the ¹H n.m.r. spectrum which showed the four olefinic protons as two groups of signals centred at δ 6.75 and 6.68 and a singlet due to H-7 at δ 3.06. The 7-substituent was converted into the lower prostaglandin side chain prior to oxidative rearrangement. Thus deprotection of (5) by treatment with tetra-n-butylammonium fluoride or HF in acetonitrile followed by reduction with lithium aluminium hydride and reprotection with t-butyldimethylsilvl chloride afforded the corresponding allyl silvl ether (6) in 67% overall yield from (5). The first synthesis of this compound from 7-formylnorbornadiene⁵ is a more difficult process and will be reported elsewhere.6

By analogy with norbornadiene which is known to give 6formylbicyclo[3.1.0]hex-2-ene (8a) via the epoxide (7a) on peracid oxidation,⁷ we hoped to obtain the bicyclic aldehyde (8b) from the *exo-anti*-epoxide (7b) intermediate. Treatment of the 7-alkenylnorbornadiene (6) with peracetic acid buffered with anhydrous sodium carbonate at 0 °C afforded the required aldehyde (8b)² which exists in equilibrium⁸ with the less polar oxabicyclo[3.2.1]octa-3,6-diene (9). Hydrolysis of the latter gave the required hydroxycyclopentenylacetaldehyde (1), in 44% yield (Scheme 2). The *endo*-epoxides [(10) and (11), 22%] were also formed as minor products during the peracetic acid oxidation.

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